



2,5-Anhydro sugar diacid and 2,5-anhydro sugar diamine based C_2 symmetric peptidomimetics as potential HIV-1 protease inhibitors

T. K. Chakraborty,^{a,*} Subhash Ghosh,^a M. H. V. Ramana Rao,^a A. C. Kunwar,^a H. Cho^b and A. K. Ghosh^b

^aIndian Institute of Chemical Technology, Hyderabad 500 007, India

^bDepartment of Chemistry, University of Illinois at Chicago, 845 West Taylor Street, Chicago, IL 60607, USA

Received 15 August 2000; revised 2 October 2000; accepted 11 October 2000

Abstract

Conformationally constrained molecular frameworks of the 2,5-anhydro sugar diacid (**9**) and 2,5-anhydro sugar diamines (**10**, **11**) were used to construct architecturally beautiful novel C_2 symmetric peptidomimetics **1–8**. Although none of these compounds showed any significant HIV-1 protease inhibitory activity, further refinements in design may lead to protease inhibitors based on these rigid carbohydrate-derived scaffolds. © 2000 Elsevier Science Ltd. All rights reserved.

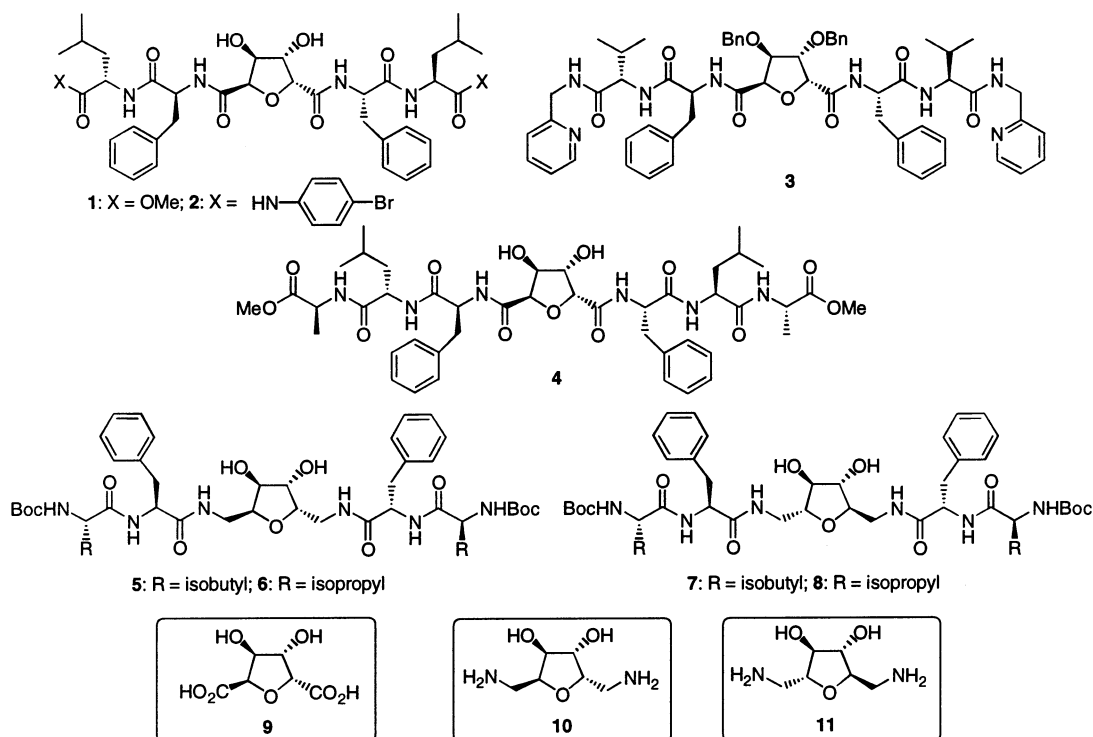
Among the several specific targets in the cell cycle of the human immunodeficiency virus (HIV), the virally encoded homodimeric HIV-1 protease, which is an aspartyl protease required for maturation of the infectious virion, has emerged as a promising target leading to the design and evaluation of a vast array of compounds with diverse structural motifs as possible inhibitors, some of which have already been approved for the treatment of AIDS.^{1–6}

We describe herein the development of a new class of compounds **1–8** as potential HIV-1 protease inhibitors that are based on carbohydrate–peptide hybrid structures. In this approach, identical peptide chains are anchored on both sides of a core carbohydrate motif—a C_2 symmetric 2,5-anhydro sugar diacid or sugar diamine—leading to the formation of C_2 symmetric peptidomimetics.⁷ Carbohydrate-based molecular designs are increasingly drawing chemists' attention.^{8,9} Detailed studies on the development of new HIV-1 protease inhibitors based on acyclic carbohydrates have recently been reported.^{10–12} It is also being increasingly felt that small molecule protease inhibitors need to have restricted degrees of freedom. The success of cyclic urea based inhibitors supports this feeling.^{13,14} This has prompted us to look for cyclic carbohydrate based core foundations as conformationally rigid scaffolds to build a new class of molecular frameworks as potential protease inhibitors.

* Corresponding author.

Two different types of cyclic carbohydrate framework—sugar diacid and sugar diamine—are employed in our designs. These molecules are 2,5-anhydro-D-idaric acid (**9**), 1,6-diamino-2,5-anhydro-1,6-dideoxy-D-iditol (**10**) and 1,6-diamino-2,5-anhydro-1,6-dideoxy-D-mannitol (**11**). The sugar diacid **9**, to the best of our knowledge, was reported for the first time by us along with the synthesis of the first sugar diacid based peptidomimetic **1**.¹⁵ The twofold symmetry of the molecule was evident in its ¹H and ¹³C NMR spectra which showed peaks for only half of the molecule. Two-dimensional NMR studies in combination with constrained molecular dynamics (MD) simulations revealed a very ordered *C*₂ symmetric structure for **1**, consisting of identical intramolecular H bonds at the two ends between the LeuNH→sugar-OH. A very small temperature coefficient for the LeuNH chemical shift ($\Delta\delta/\Delta T = -1.2$ ppb K⁻¹) was observed in DMSO-*d*₆. The presence of two ‘*cis*- β -hydroxycarboxyl’ moieties, the core structural motif believed to be responsible for such intramolecular H bonds,¹⁶ on two sides of the tetrahydrofuran ring nucleated identical β -turn-like structures in **1** at both ends.

To further these studies, we decided to prepare a few more *C*₂ symmetric peptidomimetics and examine their structures and properties, especially HIV-1 protease inhibition activities, as these compounds structurally resemble many well-established *C*₂ symmetric HIV-1 protease inhibitors. In this paper we describe the results of that investigation.

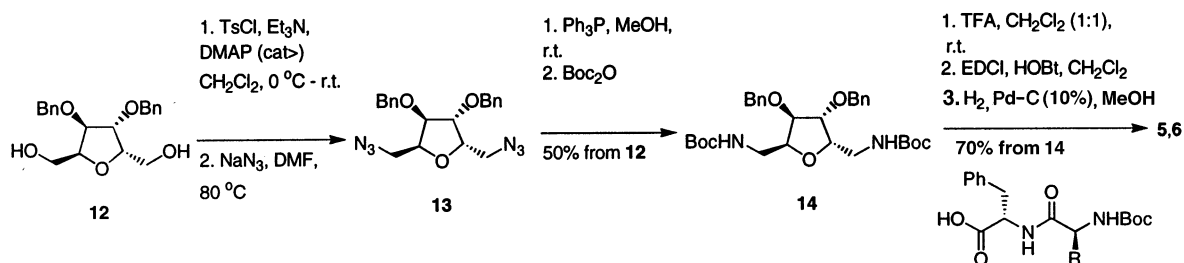


Compounds **2** and **4** were prepared in 60 and 70% yields, respectively, from **1** in two steps: saponification followed by the reaction with *p*-bromoaniline (for **2**) or H₂N-Ala-OMe (for **4**) under standard solution phase peptide synthesis conditions using 1-hydroxybenzotriazole (HOBt) and 1-ethyl-3-(3-(dimethylamino)propyl)carbodiimide hydrochloride (EDCI) as coupling agents and amine-free dry DMF and/or dry CH₂Cl₂ as solvents. The synthesis of **3** was carried out following the same procedure described earlier for the synthesis of **1**.¹⁵ The

di-*O*-benzyl derivative of 2,5-anhydro-D-idaric acid¹⁵ (**9**) was treated with an excess of amino-terminal free peptide molecule to furnish **3** in 55% yield.

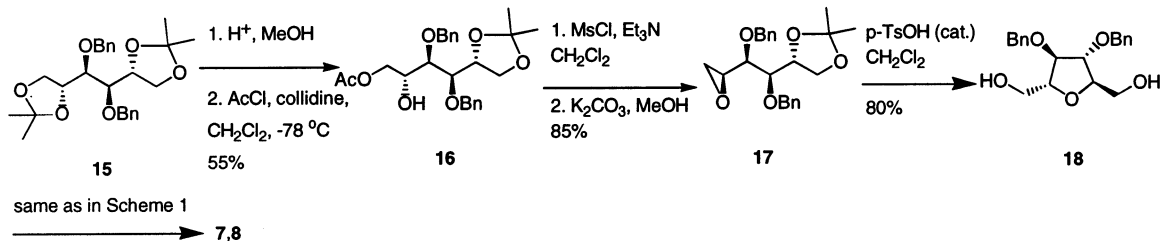
Structural studies of these molecules (**2–4**) did not reveal significant turn structures in solution as seen in **1**. While compounds **3** and **4** still maintained C_2 symmetric structures as shown by their NMR studies in DMSO- d_6 , the *p*-bromoanilide derivative **2** lost its symmetry with each of its amide protons showing two peaks in the same solvent. Its LeuNH chemical shifts, however, exhibited low temperature coefficients of -2.1 and -2.4 ppb K^{-1} maintaining their participation in intramolecular H bonding, similar to the one observed in **1**, albeit weaker. Hydrogenation of **3** gave a complex mixture of products and its structure and properties were studied in the Bn-protected form, which showed high temperature coefficients for the amide proton chemical shifts. The temperature coefficient of the LeuNH peak in **4** was -2.8 ppb K^{-1} , indicating gradual loss in its ability to nucleate the desired secondary structure, probably due to the increased steric hindrance.

Next, we turned our attention to the remaining peptidomimetics, **5–8**, based on sugar diamines **10** and **11**. Syntheses of **5** and **6** were accomplished according to Scheme 1. The starting material **12**, prepared from 1,3:4,6-di-*O*-benzylidene-D-mannitol,¹⁵ was converted to **14** in four steps—ditosylation, azidation and selective azide reduction, followed by protection in situ using Boc₂O for purification purposes giving the di-*N*-Boc-protected intermediate **14** in 50% overall yield from **12**. Purified **14** was treated with TFA in CH₂Cl₂ and peptide chains were attached at both ends using EDCI–HOBT as coupling reagents. Finally, debenzylation using H₂–Pd–C (10%) in MeOH furnished the final product **5** (and **6**) in ~70% overall yield.¹⁸



Scheme 1. Synthesis of **5** and **6**

Syntheses of **7** and **8** from 3,4-di-*O*-benzyl-1,2:5,6-diisopropylidene-D-mannitol (**15**) are depicted in Scheme 2. Monodeprotection gave a diol intermediate whose primary hydroxyl was selectively acylated to get a 55% yield of acetate **16**. Conversion of **16** to the terminal epoxide **17** was achieved in two steps in 86% yield. Acid-catalyzed cycloetherification of **17** went smoothly to furnish the 2,5-anhydro-D-mannitol framework **18**, which was converted to the



Scheme 2. Synthesis of **7** and **8**

diamine and finally to the desired products **7** and **8**,¹⁹ following the procedure described above in Scheme 1.

All the sugar diamine based mimetics, **5–8**, had C_2 symmetric structures in solution, as determined by their ^1H NMR studies in $\text{DMSO-}d_6$, which showed peaks for only half of the molecule. However, very high temperature coefficients for all of their amide proton chemical shifts indicated the absence of any intramolecular H bonding in these molecules.

The potential of the peptidomimetics **1–8** as HIV protease inhibitors were evaluated using an assay protocol developed by Toth and Marshall.²⁰ Unfortunately, none of these compounds showed any enzyme inhibitory activities up to 2 μM inhibitor concentration. The large distances between the phenyl rings of the P1/P1' positions are probably responsible for their low activities. Efforts are now underway to improve our designs and to prepare active HIV-1 protease inhibitors from conformationally constrained 2,5-anhydro sugar-based scaffolds.

Acknowledgements

Authors wish to thank the CSIR, New Delhi for research fellowships (to S.G. and M.H.V.R.R.) and a Young Scientist Award Research Grant (to T.K.C.).

References

1. Leung, D.; Abbenante, G.; Fairlie, D. P. *J. Med. Chem.* **2000**, *43*, 305–341 and references cited therein.
2. Molla, A.; Granneman, G. R.; Sun, E.; Kempf, D. J. *Antiviral Res.* **1998**, *39*, 1–23.
3. Cohen, J. *Science* **1996**, *272*, 1880–1883.
4. Vacca, J. P. In *Methods in Enzymology*, Kus, L. C.; Shafer J. A., Eds.; Academic Press: San Diego, 1994; Vol. 241, pp. 311–334.
5. Kempf, D. J. In *Methods in Enzymology*, Kus, L. C.; Shafer J. A., Eds.; Academic Press: San Diego, 1994; Vol. 241, pp. 334–354.
6. Wlodawer, A.; Erickson, J. W. *Annu. Rev. Biochem.* **1993**, *62*, 543–585.
7. For a recent article on C_2 symmetric HIV-1 protease inhibitors see: Keinan, S.; Avnir, D. *J. Am. Chem. Soc.* **2000**, *122*, 4378–4384 and references cited therein.
8. Schweizer, F.; Hindsgaul, O. *Curr. Opin. Chem. Biol.* **1999**, *3*, 291–298.
9. Drickamer, K.; Dwek, R. A. *Curr. Opin. Struct. Biol.* **1995**, *5*, 589–590.
10. Alterman, M.; Björnsne, M.; Mühlman, A.; Classon, B.; Kvarnström, I.; Danielson, H.; Markgren, P.; Nillroth, U.; Unge, T.; Hallberg, A.; Samuelsson, B. *J. Med. Chem.* **1998**, *41*, 3782–3792.
11. Zuccarello, G.; Bouzide, A.; Kvarnström, I.; Niklasson, G.; Svensson, S. C. T.; Brisander, M.; Danielsson, H.; Nillroth, U.; Karlén, A.; Hallberg, A.; Classon, B.; Samuelsson, B. *J. Org. Chem.* **1998**, *63*, 4898–4906.
12. Wachtmeister, J.; Mühlman, A.; Classon, B.; Kvarnström, I.; Hallberg, A.; Samuelsson, B. *Tetrahedron* **2000**, *56*, 3219–3225.
13. Pierce, M. E.; Harris, G. D.; Islam, Q.; Radesca, L. A.; Storace, L.; Waltermire, R. E.; Wat, E.; Jadhav, P. K.; Emmett, G. C. *J. Org. Chem.* **1996**, *61*, 444–450.
14. Lam, P. Y. S.; Jadhav, P. K.; Eyermann, C. J.; Hodge, C. N.; Ru, Y.; Bachelier, L. T.; Meek, J. L.; Otto, M. J.; Rayner, M. M.; Wong, Y. N.; Chang, C.-H.; Weber, P. C.; Jackson, D. A.; Sharpe, T. R.; Erickson-Viitanen, S. *Science* **1994**, *263*, 380–384.
15. Chakraborty, T. K.; Ghosh, S.; Jayaprakash, S.; Sharma, J. A. R. P.; Ravikanth, V.; Diwan, P. V.; Nagaraj, R.; Kunwar, A. C. *J. Org. Chem.* **2000**, *65*, 6441–6457.
16. Chakraborty, T. K.; Jayaprakash, S.; Diwan, P. V.; Nagaraj, R.; Jampani, S. R. B.; Kunwar, A. C. *J. Am. Chem. Soc.* **1998**, *120*, 12962–12963.

17. Selected physical data of **4**. $R_f=0.4$ (silica, EtOAc); $[\alpha]_D^{20} -12.8$ (*c* 0.52, CHCl₃); ¹H NMR (500 MHz, DMSO-*d*₆): δ 8.09 (d, $J=6.8$ Hz, 2H, AlaNH), 8.07 (d, $J=8.5$ Hz, 2H, PheNH), 7.72 (d, $J=8.2$ Hz, 2H, LeuNH), 7.28–7.19 (m, 10H, aromatic), 5.96 (d, $J=2.8$ Hz, 2H, OH), 4.56 (d, $J=2.8$ Hz, 2H, C2- and C5-*H*), 4.50 (ddd, $J=10.4$, 8.5, 3.4 Hz, 2H, Phe α H), 4.28 (ddd, $J=9.8$, 8.2, 5.1 Hz, 2H, Leu α H), 4.23 (m, 2H, Ala α H), 4.16 (t, $J=2.8$ Hz, 2H, C3- and C4-*H*), 3.60 (s, 6H, OCH₃), 3.19 (dd, $J=14.1$, 3.4 Hz, 2H, Phe β H), 2.96 (dd, $J=14.1$, 10.4 Hz, 2H, Phe β H'), 1.63 (m, 2H, Leu γ H), 1.44 (ddd, $J=13.7$, 8.6, 5.1 Hz, 2H, Leu β H), 1.38 (ddd, $J=13.7$, 9.8, 5.6 Hz, 2H, Leu β H'), 1.29 (d, $J=7.3$ Hz, 6H, Ala β H), 0.89 (d, $J=7.2$ Hz, 6H, Leu δ H), 0.8 (d, $J=7.2$ Hz, 6H, Leu δ H'); ¹³C NMR (125 MHz, CDCl₃): δ 172.92, 172.25, 170.51, 169.47, 135.79, 128.94, 128.91, 127.51, 83.26, 77.92, 54.19, 52.43, 51.35, 48.46, 40.39, 36.55, 24.48, 22.89, 21.72, 17.49.
18. Selected physical data of **6**. $R_f=0.3$ (silica, EtOAc); $[\alpha]_D^{20} -23.3$ (*c* 0.5, CHCl₃); ¹H NMR (500 MHz, DMSO-*d*₆): δ 8.04 (t, $J=5.6$ Hz, 2H, sugarNH), 7.85 (d, $J=8.4$ Hz, 2H, PheNH), 7.24–7.14 (m, 10H, aromatic), 6.64 (d, $J=9.0$ Hz, 2H, ValNH), 5.05 (d, $J=3.6$ Hz, 2H, OH), 4.56 (ddd, $J=9.2$, 8.4, 5.1 Hz, 2H, Phe α H), 3.91 (dt, $J=6.8$, 2.8 Hz, 2H, C2- and C5-*H*), 3.82 (dd, $J=3.6$, 2.8 Hz, 2H, C3- and C4-*H*), 3.70 (dd, $J=9.0$, 7.0 Hz, 2H, Val α H), 3.24 (ddd, $J=13.6$, 6.8, 5.8 Hz, 2H, C1- and C6-*H*), 3.12 (ddd, $J=13.6$, 6.8, 5.6 Hz, 2H, C1- and C6-*H*'), 2.94 (dd, $J=13.7$, 5.1 Hz, 2H, Phe β H), 2.79 (dd, $J=13.7$, 9.2 Hz, 2H, Phe β H'), 1.81 (m, 2H, Val β H), 1.37 (s, 18H, Boc), 0.71 (d, $J=6.6$ Hz, 6H, Val γ H), 0.67 (d, $J=6.6$ Hz, 6H, Val γ H'); ¹³C NMR (125 MHz, DMSO-*d*₆): δ 171.38, 170.87, 155.27, 137.47, 129.11, 127.91, 126.15, 78.53, 78.06, 75.84, 59.99, 53.48, 38.29, 37.92, 30.36, 28.11, 19.02, 18.05.
19. Selected physical data of **8**. $R_f=0.45$ (silica, 10% MeOH in CHCl₃); $[\alpha]_D^{20} +29.5$ (*c* 0.55, CHCl₃); ¹H NMR (500 MHz, DMSO-*d*₆): δ 8.07 (dd, $J=6.3$, 5.2 Hz, 2H, sugarNH), 7.81 (d, $J=8.7$ Hz, 2H, PheNH), 7.23–7.14 (m, 10H, aromatic), 6.65 (d, $J=9.2$ Hz, 2H, ValNH), 5.17 (d, $J=3.9$ Hz, 2H, OH), 4.60 (ddd, $J=9.7$, 8.7, 4.3 Hz, 2H, Phe α H), 3.77–3.70 (m, 4H, C2-, C3-, C4- and C5-*H*), 3.69 (dd, $J=9.2$, 6.8, 1H, Val α H), 3.27 (dt, $J=13.2$, 5.2 Hz, 2H, C1- and C6-*H*), 3.19 (dt, $J=13.2$, 6.3 Hz, 2H, C1- and C6-*H*'), 2.97 (dd, $J=13.7$, 4.3 Hz, 2H, Phe β H), 2.77 (dd, $J=13.7$, 9.7 Hz, 2H, Phe β H'), 1.79 (m, 2H, Val β H), 1.37 (s, 18H, Boc), 0.69 (d, $J=6.8$ Hz, 6H, Val γ H), 0.64 (d, $J=6.8$ Hz, 6H, Val γ H'); ¹³C NMR (125 MHz, DMSO-*d*₆): δ 171.10, 170.86, 155.27, 137.64, 129.18, 127.87, 126.11, 82.12, 78.81, 78.08, 60.10, 53.41, 41.10, 38.13, 30.41, 28.13, 19.04, 18.08.
20. Toth, M. V.; Marshall, G. R. *Int. J. Pep. Prot. Res.* **1990**, *36*, 544–550.